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CONSTANT Calculate the molar gas
constant R in the following units: a.
(atm)(cm³)/(g·mol)(K) b.

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Calculation Procedure 1. Assume a basis.
Assume gas is at standard conditions,
that is, 1 g·mol gas at 1 atm (101.3 kPa)
pressure and 0 C (273 K, or 492 R),
occupying a volume of 22.4 L. 2.
Compute the gas constant. Apply
suitable conversion factors and obtain

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the gas constant in various units. Use $PV = RT$; that is, $R = PV/T$. Thus, a.

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